Remediation of Low Level Hexavalent Chromium from Water by Activated Sludge Process: A Review

Aliya Naz¹, Gupta, S.K.²

¹Junior Research Fellow, Department of Environmental Science and Engineering, ISM Dhanbad-826004, Jharkhand, India. ²Department of Environmental Science and Engineering, ISM Dhanbad-826004, Jharkhand, India

Abstract: The techniques available for the treatment of hexavalent chromium [Cr (VI)] include, chemical reduction followed by precipitation, ion exchange, membrane separation, and reverse osmosis. However, they are not techno-economically feasible due to high cost and problems of disposal of low density sludge generated in these processes. Microbial remediation techniques i.e. both aerobic and anaerobic processes have shown enough potential using single consortia in lab scale reactor and researchers have advocated that this can prove to be the most economical technique. Activated sludge process is the most common treatment system for municipal and industrial wastewater. However, high concentration of Cr (VI) (>5ppm) can affect the microbial growth of activated sludge. Concentration of Cr (VI) at which it stimulates or affects the growth of microorganisms is often controversial. The aim of this review is to investigate the efficiency of activated sludge process for the remediation of low concentration of Cr (VI) (<2ppm) from water. Most of the studies revealed that microbial reduction followed by precipitation and biosorption could be one of the possible mechanisms for the removal of Cr (VI). Instead of physiochemical treatment adopted by mine industries in India, Activated sludge process can efficiently be used for the removal of low concentration of Cr (VI).

Keywords: Hexavalent Chromium, Activated Sludge Process, Biosorption, Precipitation

1. INTRODUCTION

Chromium is a heavy metal used for the various purposes in many industries such as steel alloys, chemical manufacturing, paint and dyes, tanneries etc. This is usually released in the environment in the form of Cr (VI) and Cr (III) state. Both the form of chromium is found injurious however Cr (VI) is 100 times more harmful than Cr (III) even at the low concentration [1].

The Sukinda Vally of Odisha accounts for 93% of total chromite production in India [2]. The Cr (VI) generation is the major environmental concern of this region. The contamination of Cr (VI) in local water bodies and underground water is the most serious problem due to its carcinogenic properties. According to World health Organization (WHO-2012) the level of Cr (VI) should not increase more than 0.05mg/l. Indian Standard also prescribe the maximum limit of Total Cr is

0.05mg/l in drinking water (IS 10500-2012). Maximum tolerance limit of Cr (VI) in inland water is 0.1 mg/l [3].

The most common technique used in Sukinda Valley for Cr (VI) removal is physiochemical reduction with ferrous sulfate and precipitation by lime. This process consumes huge amount of chemicals and produce a lot of low density toxic sludge which creates disposal problems. So this process is not economically feasible. There were many studies carried out on the bioremediation of Cr (VI) and researchers advocated that this may prove to be environmental friendly and feasible [4][5][6]. A number of bacterial consortia have been proved 100% efficient for the remediation of Cr (VI) from water [7]. Although Cr (VI) is very toxic and cause harm to the microorganisms' growth, a numbers of reports support that Cr (VI) does not cause harm to the microorganisms at low level (less than 5 mg/l). Reports also support that at higher level (> 25 mg/l) of Cr (VI) the activities of microorganisms in activated sludge adversely affected [8].

The aim of this review is to discuss the efficiency of ASP process for remediation of low concentration of Cr (VI). ASP is the most popular biological treatment process of wastewater. This is based on the bacterial and other microorganism population in an aerated tank. Since the process of single culture of microorganism is difficult for the large scale field, mixed cultured is easy and will give fruitful results on the large volume of wastewater with low concentration of chromium content.

2. ACTIVATED SLUDGE PROCESS FOR REMOVAL OF CR (VI):

ASPs are the most popular technique commonly used for the treatment of municipal and industrial wastewaters. There are numbers of studies have done on the metal removal efficiency of ASP reactor. ASPs are totally based on the microbial activities. Cell biomass (alive and dead both) both have property to reduce and adsorb the metal [9]. Cr (VI) is very toxic than Cr (III) and can damage the microbial cell, being less toxic Cr (III) can be more efficiently removed by ASP. Cr (VI) also can be removed through ASP in which Cr (VI) first reduced into Cr (III) by the bacterial enzymatic activities and Cr (III) settled in the sludge/ dead bacterial biomass in the form of Cr (OH)₃. The studies showed that 96-99 % of the chromium present in the ASP sludge as Cr (III) [10]. Imai and Gloyna (1988) find that at higher pH and longer retention time the adsorption capacity of sludge increase [11].

The mechanism of Cr (VI) removal involves following processes:

- Reduction of Cr (VI) to Cr (III)
- Precipitation of Cr (III)

- Adsorption of Cr (VI) and Cr (III) by suspended solids present in the system
- Biosorption by Bacterial cell
- Effect of chromium on cell biomass

The reduction of Cr(VI) to Cr(III) by different microbial culture is possible in the both aerobic and anaerobic condition. Table 1 shows different microbial species found efficient in reducing of Cr(VI) in aerobic conditions [12][13][14]. When Cr(VI) is reduced to Cr(III) due to its low solubility and impermeability to cell membrane this is either precipitated in the form Cr(OH)₃ or gets biosorbed. Although ASP works in the presence of oxygen, studies show that the oxidation of Cr(III) to Cr(VI) does not take place [11][15]. This can be attributed Cr(III) is more readily adsorbed before the oxidation reaction is facilitated [11]. Many studies also demonstrated that suspended solids concentration in the ASP greatly influences the chromium adsorption [15]. This is due to the fact that suspended solids present in the water provide more surface area for adsorption of Chromium. In the some studies large part of Cr(VI) was adsorbed by the MLSS of ASP system and reduced Cr(III) was immediately detected on the suspended solids [10]. Both forms of Chromium are adsorbed by the suspended solids, although some studies says there are no any correlation between Cr(VI) and suspended solids[15].

Name of microbes	Mechanism	Description	Reference
Psudomonas	Bioreduction	Cr(VI) removal 70-99.7%	[16]
fluorecens(LB300)			
Psudomonas ambigua G-1	Bioreduction	At HRT 36 hours it reduces 76.6%	[17]
		of Cr(VI).	
Bacillus subtilis	Biosorption	Reduce Cr(VI) in range of 0.1 to	[18]
		1mM.conc.	
Acinetobactor	Bioreduction	Complete Cr(VI) reduction occur	[19]
haemolyticus		up to 10-30 mg/L. Incomplete	
		reduction takes place at >70 mg/L.	
Bacillus sphaericus	Bioreduction	Tolerate 800 mg/L Cr(VI) and	[20]
		reduce $>80\%$ of Cr(VI).	
Streptomyces griseus	Bioreduction	Can reduce 5–60 ppm of Cr(VI)	[21]
B. cereus ATCC 10987,	Bioreduction	100% reduction of Cr(VI) in 65h	[22]
B.cereus 213 16S,		incubation.	
B.Thueningenesis,			
B.myocides Microbacterium			

Table 1: Microbial species capable of reducing Cr(VI)

3. DISCUSSION

There are several possible mechanisms in the remediation of Cr(VI) from ASP as describe above. The given Cr(VI) dose in the ASP system might be reduced into Cr(III) by suspended particles present in the ASP reactor and biosorbed or precipitated through the bacterial cell wall [15]. Precipitated metals may be removed either by independent settling or by physical trapping in the sludge floc matrix. Stasinakis (2004) removed 40% of the total added Cr(VI) during activated sludge process and the remaining Cr(VI) was sorbed by the sludge flocs mainly in the form of Cr(III)[10]. According to Chen 2005 an acclimated activated sludge remove 100% of Cr(VI) and approx 98.56% total chromium at influent level 20mg/day. However, at higher concentration (60 mg/day) of Cr(VI) the removal efficiency decreases. The increase of glucose concentration in influent improves the efficiency i.e. increased from 98.64 to 100% for Cr(VI) and from 97.16 to 98.48% for Cr(III) [23].

Several studies have been done on the effect of Chromium on the microorganisms of ASP [24][8][25]. However Cr(VI) is toxic and influence the growth of microorganisms in ASP and most of researchers refer that small concentration (1- 5mg/l) of Cr(VI) prove beneficial microorganisms and stimulate the growth of microorganisms in ASP [25].Some studies refer that after an optimum Cr(VI) concentration, the reduction rate gradually decreases and the microbes started to loss their reducing efficiency due to Cr(VI) toxicity [26].The maximum dose at which the microorganisms significantly affected and reduce their reducing capacity is often controversial. Some researchers proven that the presence of Cr(VI) in concentration range of 10-50mg/l has no any harmful effect on the growth of microorganism in ASP [24][8].

4. CONCLUSION

Conventional methods used for low level of chromium is techno-economically not feasible due to operational and maintenance cost in the form of chemical and disposal of huge volume of less density toxic sludge. Reduction of Cr (VI) can be achieved > 95% by ASP system for the wastewater containing low levels of Cr (VI). Removal of Cr (VI) is mediated by microbial reduction followed by precipitation and biosorption of Cr (III). Presence of SS helps and stimulates the adsorption of reduced Cr (III). On the economical point of view ASP treatment system could be prove more efficient and economically cheap than conventional physiochemical treatment systems currently used in various Chromite Mines in India.

REFERENCES

- [1] Katz S. A, Salem H. The toxicology of chromium with respect to its chemical speciation: A review. Jornal of Applied Toxicologym1993;13(3):217-224.
- [2] IBM (Indian bureau of Mines) Indian Mineral Year book 2012.
- [3] Indian Standard; IS: 2490, Part-I 1981.

- [4] Samuel J, Paul M.L, Pulimi M. J, Chandrasekaran N, Mukherjee A. *Hexavalent chromium bioremoval through Adsorption and Consortia development from sukinda chromite mines isolates*. Industrial and Engineering Chemistry Research; 2012.
- [5] Jianlong, W., Zeyu, M., Xuan, Z., 2004. *Response of Sccharomyces cerevisiae to chromium stress*. Process Biochem. 39, 1231-1235.
- [6] Srinath T, Verma T, Ramteke P, Garg S.K. *Chromium (VI) biosorption and bioaccumulation by chromate resistant bacteria.* Chemosphere 2002; (48) 427-435.
- [7] Kamaudeen S, Arunkumar K, Avudainaygam S, Ramasamy K. *Bioremediation of chromium contaminated Environment*. Indian Journal of Experimental Biology 2003;41:972-985.
- [8] Moore W.A, McDermott G.W., Post M.A, Mendia J.W, Ettinger F. *Effects of chromium on the activated sludge process.* J. Wat. Pollut. Control Fed. 1961; 33, 54-60.
- [9] Malik A. "*Metal bioremediation through growing cells*". Environment International 2004; (30) 261 278.
- [10] Stasinakis A.S, Thomaidis N.S, Mamais D, Karivali, Lekkas T.D. *Chromium species behavior in the activated sludge process.* Chemosphere 2003;(52)1059-1067.
- [11] Imai A, Gloyna E. F, *The behavior of chromium in the activated sludge process*. Tech. Rep 1988. (226):188.
- [12] Aguilera S, Aguilar, M E, Cha´ vez M P, Lo´ pez-Meza J E, Pedraza-Reyes M, Campos-Garcı´a, J, Cervantes, C. Essential residues in the chromate transporter ChrA of Pseudomonas aeruginosa. FEMS Microbiology Letters 2004; (232):107–112.
- [13] Vaneechoutte M, Ka^m mpfer P, Baere T.D, Falsen E, Verschraegen G. Wautersia gen. nov. a novel genus accommodating the phylogenetic lineage including Ralstonia eutropha and related species, and proposal of Ralsonia [Pseudomonas] syzygii (Roberts et al. 1990) comb. nov. International Journal of Systematic and Evolutionary Microbiology 2004; (54):317–327.
- [14] McLean J.S, Beveridge T.J, Phipps D. Isolation and characterization of a chromium-reducing bacterium from a chromated copper arsenate-contaminated site. Environmental Microbiology 2000;(2):611–619.
- [15] Stasinakis A.S, Thomaidis N.S, Mamais D, Karivali M, Themistokles Lakkas D. *Chromium species* behavior in the activated sludge process. Chemosphere 2003; 52: 1059-1067.
- [16] DeLeo P.C, Ehrlich H. L. *Reduction of hexavalent chromium by pseudomonas fluorescens LB300 in batch and continuous culture*. Applied Microbiology and Biotechnology 1994;40 (5):756-759.
- [17] Losi M E, Amrhein C, Frankenberg W T, *Environmental biochemistry of chromium*. Rev Environ Contam Toxicol 1994; 136: 92.
- [18] Garbisu C, Alkorta I, Llama M.J,Serra J.L. Aerobic chromium reduction by Bacillus subtilis.Biodegradation 1998;9(2):133-141.
- [19] Zakariaa Z.A, Zakariaa Z, Surifb S, Ahmada W.A. *Hexavalent chromium reduction by Acinetobacter haemolyticus isolated from heavy-metal contaminated wastewater*. Journal of Hazardous Materials 2007;(146):30–38.
- [20] Pal A, Paul A.K. Aerobic chromate reduction by chromium-resistant bacteria isolated from serpentine soil. Microbiological Research 2004;159(4):347-354.
- [21] Laxman R.S, More S. *Reduction of hexavalent chromium by Streptomyces griseus*. Minerals Engineering 2002; (15) : 831–837.
- [22] Molokwane P.E., Meli K.C., Nkhalambayausi-Chirwa E.M. Chromium (VI) reduction in activated sludge bacteria exposed to high chromium loading: Brits culture (South Africa). water research 2008; (42):4538–4548.

- [23] Chen Y, Gu G. Preliminary studies on continuous chromium(VI) biological removal from wastewater by anaerobic-aerobic activated sludge process. Biorecourse 2005;96:1713-1721.
- [24] Barth E.F, Ettinger M.B, Salotto B.V, Mc Dermott G.N. *Summary report on the effect of heavy metals on the biological treatment process*. Journal of Water Pollution Control Federation 1965; 37:496-502.
- [25] Gokcay C.F, Yetis U. *Effect of chromium (VI) on activated sludge*. Water research 1991;25 (1): 65-73.
- [26] Chirwa, E. N., Wang Y. T. Simultaneous chromium(VI) reduction ad phenol degradation in an anaerobic consortium of bacteria. Water Res. 2000; 34: 2376-2384.